

An Approach to the Bis-oxazole  
Macrocycle of Diazonamides

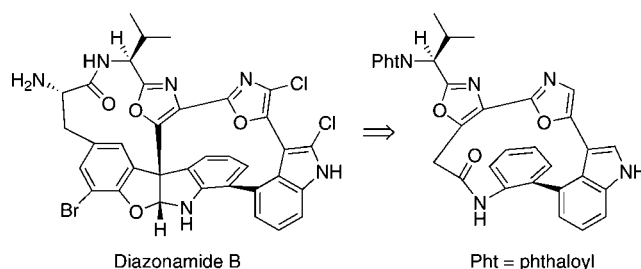
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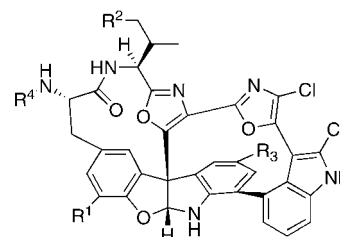
## ABSTRACT



A convergent approach to a macrocyclic compound embodying the complete “eastern” quadrant of diazonamides is described. The opening sequence in this work relies on an oxazole-forming reaction devised earlier in this group, while a late step involves a Robinson–Gabriel cyclization of an amidoketone to form a second oxazole.

Few natural products have captured so much attention in the synthetic arena as the diazonamides, **1** (Scheme 1).<sup>1</sup> Indeed, intense activity aiming to conquer their delightfully intricate structure, initially engendered by the discovery of the first members of the family,<sup>1</sup> continues unabated to this day.<sup>2</sup> To wit, since the successful conclusion of the initial synthetic campaign<sup>3</sup> and relative background work,<sup>4</sup> two formal syntheses of **1a**<sup>5</sup> have appeared in the literature, as have about half<sup>6</sup> of the numerous synthetic studies published to date.<sup>7</sup> Noteworthy biological properties<sup>8</sup> add a significant dimension to such chemical pursuits, which, understandably, have largely focused on the creation of the problematic stereogenic

Scheme 1. Structures of Diazonamides A–E



- 1a** diazonamide A  $R^1 = R^2 = R^3 = H$ ;  
 $R^4 = (S)$ -2-hydroxy-3-methylbutanoyl  
**1b** diazonamide B  $R^1 = Br$ ,  $R^2 = R^3 = R^4 = H$   
**1c** diazonamide C  $R^1 = R^2 = R^3 = H$ ;  
 $R^4 = (S)$ -2-amino-3-methylbutanoyl  
**1d** diazonamide D  $R^1 = R^4 = H$ ,  $R^2 = Me$ ,  $R^3 = Cl$   
**1e** diazonamide E  $R^1 = R^3 = R^4 = H$ ;  $R^2 = Me$

quaternary center at the core of the molecules. Yet, the assembly of the bis-oxazole-containing macrocycle that constitutes the “eastern” portion of diazonamides also raises interesting chemical issues.<sup>7</sup> In that connection, a recent report from the Moody group<sup>9</sup> describes the remarkable

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(2) Reviews: (a) Ritter, T.; Carreira, E. M. *Angew. Chem., Int. Ed.* **2002**, *41*, 2489. (b) Lachia, M.; Moody, C. J. *Nat. Prod. Rep.* **2008**, *25*, 227. See also: (c) Nicolaou, K. C.; Chen, J. S. *Pure Appl. Chem.* **2008**, *80*, 727. (d) Delle Monache, G.; Misiti, D.; Zappia, G. *Critical surveys covering the year 2002: total synthesis of natural products*. In: *Seminars in Organic Synthesis, 28th Summer School “A. Corbella”, Gargnano, Italy, June 16–20, 2003*; Marcantoni, E.; Renzi, G., Eds.; Societa di Chimica Italiana: Rome, Italy, 2003; pp 329–352.

biomimetic conversion of 2–3 into 4–5, respectively, under oxidative conditions (Scheme 2). Such a development

(3) Synthesis of putative diazonamide and structural revision: (a) Li, J.; Jeong, S.; Esser, L.; Harran, P. G. *Angew. Chem., Int. Ed.* **2001**, *40*, 4765. (b) Li, J.; Burgett, A. W. G.; Esser, L.; Amezcua, C.; Harran, P. G. *Angew. Chem., Int. Ed.* **2001**, *40*, 4770. Synthesis of actual diazonamide and structural confirmation: (c) Nicolaou, K. C.; Bella, M.; Chen, D. Y.-K.; Huang, X.; Ling, T.; Snyder, S. A. *Angew. Chem., Int. Ed.* **2002**, *41*, 3495. (d) Nicolaou, K. C.; Rao, P. B.; Hao, J.; Reddy, M. V.; Rassias, G.; Huang, X.; Chen, D. Y.-K.; Snyder, S. A. *Angew. Chem., Int. Ed.* **2003**, *42*, 1753. (e) Nicolaou, K. C.; Hao, J.; Reddy, M. V.; Rao, P. B.; Rassias, G.; Snyder, S. A.; Huang, X.; Chen, D. Y.-K.; Brenzovich, W. E.; Giuseppone, N.; Giannakakou, P.; O'Brate, A. J. *Am. Chem. Soc.* **2004**, *126*, 12897. (f) Nicolaou, K. C.; Chen, D. Y.-K.; Huang, X.; Ling, T.; Bella, M.; Snyder, S. A. *J. Am. Chem. Soc.* **2004**, *126*, 12888. (g) Nicolaou, K. C.; Snyder, S. A.; Giuseppone, N.; Huang, X.; Bella, M.; Reddy, M. V.; Rao, P. B.; Koumbis, A. E.; Giannakakou, P.; O'Brate, A. J. *Am. Chem. Soc.* **2004**, *126*, 10174. (h) Nicolaou, K. C.; Snyder, S. A.; Huang, X.; Simonsen, K. B.; Koumbis, A. E.; Bigot, A. J. *Am. Chem. Soc.* **2004**, *126*, 10162. (i) Burgett, A. W. G.; Li, Q.; Wei, Q.; Harran, P. G. *Angew. Chem., Int. Ed.* **2003**, *42*, 4961.

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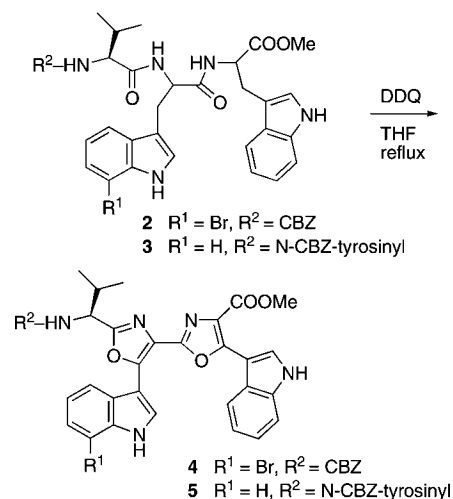
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(7) Synthetic studies up to 2002: (a) Feldman, K. S.; Eastman, K. J.; Lessene, G. *Org. Lett.* **2002**, *4*, 3525. (b) Konopelski, J. P.; Hottenroth, J. M.; Oltra, H. M.; Veliz, E. A.; Yang, Z. C. *Synlett* **1996**, 609. (c) Hang, H. C.; Drotloff, E.; Elliott, G. I.; Ritsema, T. A.; Konopelski, J. P. *Synthesis* **1999**, 398. (d) Radspieler, A.; Liebscher, J. *Synthesis* **2001**, 745. (e) Schley, D.; Radspieler, A.; Christoph, G.; Liebscher, J. *Eur. J. Org. Chem.* **2002**, 369. (f) Magnus, P.; Kreisberg, J. D. *Tetrahedron Lett.* **1999**, *40*, 451. (g) Magnus, P.; McIver, E. G. *Tetrahedron Lett.* **2000**, *41*, 831. (h) Kreisberg, J. D.; Magnus, P.; McIver, E. G. *Tetrahedron Lett.* **2001**, *42*, 627. (i) Chan, F.; Magnus, P.; McIver, E. G. *Tetrahedron Lett.* **2000**, *41*, 835. (j) Magnus, P.; Lescop, C. *Tetrahedron Lett.* **2001**, *42*, 7193. (k) Moody, C. J.; Doyle, K. J.; Elliott, M. C.; Mowlem, T. J. *Pure Appl. Chem.* **1994**, *66*, 2107. (l) Moody, C. J.; Doyle, K. J.; Elliott, M. C.; Mowlem, T. J. *J. Chem. Soc., Perkin Trans. 1* **1997**, *16*, 2413. (m) Lach, F.; Moody, C. J. *Tetrahedron Lett.* **2000**, *41*, 6893. (n) Bagley, M. C.; Hind, S. L.; Moody, C. J. *Tetrahedron Lett.* **2000**, *41*, 6897. (o) Bagley, M. C.; Moody, C. J.; Pepper, A. G. *Tetrahedron Lett.* **2000**, *41*, 6901. (p) Boto, A.; Ling, M.; Meek, G.; Pattenden, G. *Tetrahedron Lett.* **1998**, *39*, 8167. (q) Vedejs, E.; Wang, J. *Org. Lett.* **2000**, *2*, 1031. (r) Vedejs, E.; Barda, D. A. *Org. Lett.* **2000**, *2*, 1033. (s) Vedejs, E.; Zajac, M. A. *Org. Lett.* **2001**, *3*, 2451. (t) Wipf, P.; Methot, J.-L. *Org. Lett.* **2001**, *3*, 1261. (u) Fuerst, D. E.; Stoltz, B. M.; Wood, J. L. *Org. Lett.* **2000**, *2*, 3521. See also: (v) Motallebi, S.; Mueller, P. *Helv. Chim. Acta* **1993**, *76*, 2803.

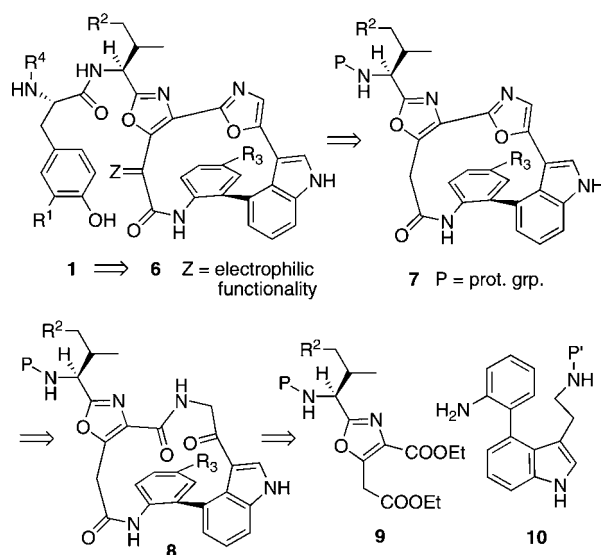
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## Scheme 2. Moody Biomimetic Oxazole Formation



prompts us to disclose results of our own efforts in this area. As outlined in Scheme 3, we have been interested in an

## Scheme 3. Approach to Diazonamides



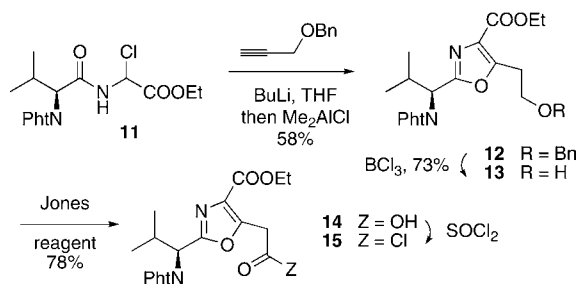
approach that would advance synthetic intermediate 6 (Z = appropriate electrophilic functionality) to the natural products. In turn, compound 6 would be obtained from 8 via Robinson–Gabriel oxazole formation and further elaboration of the resultant macrolactam 7. The synthesis of 8 could proceed through the union of oxazoles 9, which should be available through an oxazole-forming reaction developed earlier in this group,<sup>10</sup> and 4-aryltryptamine 10. Herein, we detail procedures for the implementation of such a plan.

The condensation of valine-derived chloroglycinate 11<sup>11</sup> with the dimethylaluminum acetylide prepared from

(9) Sperry, J.; Moody, C. J. *Tetrahedron* **2010**, *66*, 6483. See also ref 6g.

benzyl propargyl ether (Scheme 4; Pht = phthalimido) afforded enantiopure oxazole **12**. Without extensive

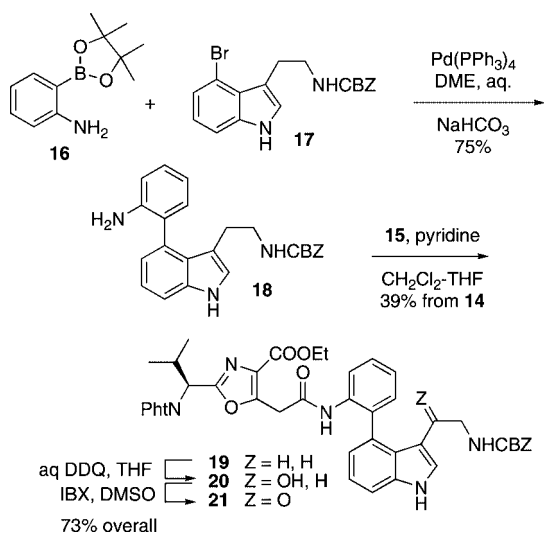
**Scheme 4.** Oxazole Formation from  $\alpha$ -Chloroglycinates



purification, this material was debenzylated ( $\text{BCl}_3$ ) to furnish alcohol **13**, a substance that proved to be quite sturdy, being best advanced to acid **14** by Jones oxidation. Acid chloride **15** emerged uneventfully upon treatment of **10** with  $\text{SOCl}_2$ .

Parallel work yielded 4-arylidole **18**, prepared by Pd-mediated coupling of commercial boronic ester **16** with the known **17**<sup>6g</sup> (Scheme 5). Substance **18** is axially chiral,

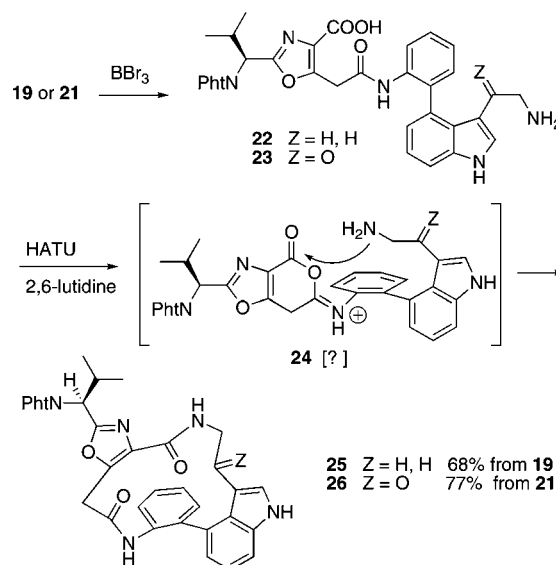
**Scheme 5.** Coupling of **16** and **18** and Oxidation of Product **19**



but for the purpose of the present study it was employed as the racemate.<sup>12</sup> The condensation of **15** with **18** was difficult, presumably on account of the hindered nature of the amino group, and it proceeded to furnish **19** in a moderate 39% yield. This material, resulting through the union of enantiopure **15** with racemic **18**, was obtained as a 1:1 mixture of atrop-diastereomers, as were all subsequent synthetic intermediates derived from it. Conversion of **19** into **21**, as required for the ultimate formation of the second oxazole ring in a Robinson–Gabriel mode, was achieved in 73% yield by hydroxylation with

DDQ, followed by IBX oxidation of the intermediate carbinol **20**.<sup>6g</sup> Both **19** and **21** underwent simultaneous cleavage of the CBZ and ethyl ester groups upon exposure to  $\text{BBr}_3$  to afford the extremely polar amino acids **22** and **23**, respectively (Scheme 6). These were not thoroughly

**Scheme 6.** Preparation of Macrocyclic Compounds **25** and **26**



characterized. Instead, they were immediately cyclized ( $\text{HATU}$ )<sup>13</sup> to afford macrolactams **25** and **26** (68% and 77% overall yield, respectively). It is unclear at this time whether this transformation occurred by direct nucleophilic attack of the amino group onto the  $\text{HATU}$ -activated carboxyl group or through the intervention of an azlactone-type intermediate such as **24**. Regardless, it is worthy of note that, contrary to the case of **19**, the action of aqueous DDQ upon **25** failed to induce hydroxylation in an efficient manner, preventing a possible conversion into **26**. In this respect, the behavior of **25** differs from that of a related system described by Moody,<sup>6g,14</sup> which underwent the oxidation in question in 84% yield, underscoring the fact that the reactivity of such macrocycles is quite sensitive to structural details.

Finally, exposure of **26** to the action of  $\text{TsOH}$  in refluxing toluene triggered formation of bis-oxazole **27**, a 1:1 mixture

(10) (a) Coqueron, P. Y.; Didier, C.; Ciufolini, M. A. *Angew. Chem., Int. Ed.* **2003**, *42*, 1411. Recent developments: (b) Zhang, J.; Coqueron, P. Y.; Vors, J. P.; Ciufolini, M. A. *Org. Lett.* **2010**, *12*, 3942. (c) Zhang, J.; Ciufolini, M. A. *Org. Lett.* **2009**, *11*, 2389. (d) Zhang, J.; Polishchuk, E. A.; Chen, J.; Ciufolini, M. A. *J. Org. Chem.* **2009**, *74*, 9140. (e) Chau, J.; Zhang, J.; Ciufolini, M. A. *Tetrahedron Lett.* **2009**, *50*, 6163. Review: (f) Zhang, J.; Coqueron, P. Y.; Ciufolini, M. A. *Heterocycles* **2010**, *77*, in press [DOI: 10.3987/REV-10-SR(E)3].

(11) Ref 10 as well as: Zhang, J. *Tetrahedron Lett.* **2010**, *51*, 4699.

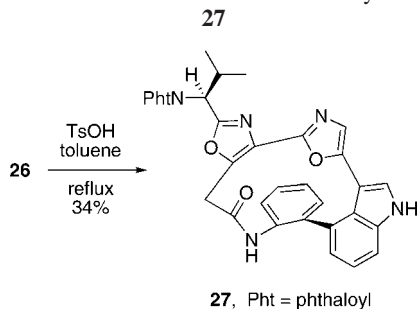
(12) No literature record of compound **18** appears to exist, but the corresponding 4-(3-aminophenyl) indole has been described in a recent patent application: Wells, J.; Renslo, A. R.; Wolan, D.; Zorn, J. PCT Int. Appl. (2009), WO 2009089508.

(13) Carpino, L. *J. Am. Chem. Soc.* **1993**, *115*, 4397.

(14) The Moody substrate differs from **25** in that the oxazole ring and the benzene nucleus are bridged by a plain methylene group (i.e., the CO–NH linkage is missing), which furthermore connects to the *meta* position of the phenyl group.

of atrop-diastereomers, in 34% yield after purification (Scheme 7). This compound embodies a specific example of generic intermediate **7**. Its successful formation in the

**Scheme 7.** Robinson–Gabriel Route to Macrocyclic Bis-oxazole



fashion just described validates the surmise of Scheme 3, it complements the Moody approach, and it delineates a heretofore unexplored route to the “eastern” segment of diazonamides. Research aiming to advance **27** to a totally synthetic diazonamide continues, and results in this area will be described in due course.

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**Supporting Information Available:** Experimental procedures and characterization data for new compounds, plus NMR ( $^1\text{H}$  and  $^{13}\text{C}$ ) spectra of several products. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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